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going from a set of reactants to a set of products, the change in enthalpy is the same whether the reaction takes place in one step or in a series of steps. energy. the ability to do work or produce heat. matter spread.

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Chemistry: Matter and
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161 Section 10.1

Measuring Matter page

320–324 Practice

Problems pages

323–324 1. Zinc (Zn) is
used to form a

corrosion-inhibiting
surface on galvanized

steel. Determine the
number of Zn atoms in

2.50 mol of Zn. 2.50

mol Zn

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CHAPTER

Understanding Main Ideas (Part B) Date

Class CHAPTER

ASSESSMENT Circle the letter of the choice that

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best completes the statement or answers the question. Use the following figure to answer questions 1 and 2. 1. According to Bohr's atomic model, which letter(s) in the figure represents a place where an electron cannot be? b. B, C and E c.

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320 Chapter 10 • The Mole Section 110.10.1 Measuring Matter MAIN Idea Chemists use the mole to count atoms, molecules, ions, and formula units. Real-World Reading Link Has your class ever had a contest to guess how

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Answer...

Modern Chemistry 91
Chapter Test Name
Class Date Chapter
Test B, continued 27.
Distinguish between
ionic crystals and
metallic crystals. PART
V Write the answers to
the following questions
on the line to the left,
and show your work in
the space provided.
The molar enthalpy of
fusion for water is
6.008 kJ/mol. 28.

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To help you prepare for the End-of-Chemistry Assessment, review the following at the end of each chapter: Chapter Highlights. Vocabulary . Answer Chapter Review questions (answers will be posted on my website next week)

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CHEMISTRY: Chapter

10 Prep-Test Matching

Match each item with

the correct statement

below. a. calorimeter d.

temperature b. calorie

e. specific heat c. joule

f. heat ____ 1. quantity

of heat needed to raise

the temperature of 1 g

of water by 1 C ____ 2.

SI unit of energy ____ 3.

energy transferred

between 2 objects

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Assessment
because of
temperature difference
Answers