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Chapter 16

Modern Chemistry 111 Chapter Test Name Class Date Chapter Test B, continued 18. Dissolution processes with negative enthalpies of solution are processes. 19. is the solution process with water as the solvent. 20. Liquid solutes and solvents that are not soluble in each other are. 21. A solute molecule that is surrounded by solvent molecules is ...

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Chemistry Chapter 16 - Solutions. HCEC High School. STUDY. PLAY. 3 dissolving factors. stirring, temperature, and surface area. saturated solution. maximum amount of solution in solvent. solubility, grams of solute per 100 grams of solvent. unsaturated solution. less solution than a saturated solution.

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Chemistry test - chapter 16: solutions. STUDY. PLAY. Soluion. A homogenous mixture. Solute. Dissolved particles in a solution. Solvent. The dissolving medium in a solution. Solubility. The amount of a substance that dissolves in a given quantity of solvent at specified conditions of T and P to produce a saturated solution.

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A.P. Chemistry Practice Test: Ch. 11, Solutions Name _____ MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) Formation of solutions where the process is endothermic can be spontaneous provided that _____. A)the solvent is a gas and the solute is a solid

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Chapter 16 Sample Test Instructions: You may use a "clean" periodic table. Sample test questions for the optional sections on molality, normality, and colligative properties follow this test. 1) A solution that can dissolve more of a substance than it currently contains is said to be ____.

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05 CTR ch16 7/12/04 8:14 AM Page 399 PROPERTIES OF ...

Chapter 16 Solutions 16.1 Properties of Solutions 16.2 Concentrations of Solutions 16.3 Colligative Properties of Solutions 16.4 Calculations Involving ... CHEMISTRY & YOU How do you think crystal-growing kits work? Use what you know about solubility and supersaturated solutions

CHEMISTRY: SOLUTIONS MULTIPLE CHOICE TEST

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