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SECTION 22.1 HYDROCARBONS(pages 693-701)

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ch. 22 Hydrocarbon Compounds. Pearson Success Chapter 22. STUDY. PLAY. hydrocarbons. the simplest organic compounds containing only carbon and hydrogen. alkane. a hydrocarbon in which there are only single covalent bonds. straight-chain alkane. contain any number of carbon atoms, one after the other, in a chain; ethane is simplest.

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Hydrocarbon compounds Section 22.1: Hydrocarbons What is the relationship between carbon and organic chemistry? 200 years ago scientists thought that only living organisms could make carbon compounds known in living cells Organic means related to or made by living things In 1828, A

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German Chemist, Friedrich Wohler

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Chapter 22 Hydrocarbon Compounds 239 SECTION 22.2 UNSATURATED HYDROCARBONS (pages 702–703) This section explains the difference between unsaturated and saturated hydrocarbons. It also describes the difference between alkenes and alkynes. Alkenes (page 702) 1. What is an alkene? 2. Organic compounds that contain the maximum number of hydrogen ...

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Hydrocarbon Compounds 719 CHAPTER 22 Assessment 37. pentane: $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ hexane: $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ 38. a. propane b. octane c. pentane 39. 40. a. 2-methylbutane b. 2,3-dimethylbutane c. 3-methylhexane 41. The carbon-carbon bonds are non-polar and the carbon-hydrogen

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This section describes the bonding in hydrocarbons and ...

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Chapter 21: Hydrocarbons

You are at chapter 22!! Let's goooo!!!! First of all, a hydrocarbon is the simplest organic compound and it only contains carbon and hydrogen. For this, we can use our background knowledge to know that since carbon has four valence electrons, it can form four covalent bonds.

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