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## **SECTION 25.1 NUCLEAR RADIATION (pages 799-802)**

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Types of Radiation

Discuss Explain that

the nuclei of a

radioactive element

spontaneously

decompose. Nuclear

chemistry is the study

of changes in matter

that originate in atomic

nuclei.

**PowerPoint Chapter  
18: Nuclear**

*Page 7/28*

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## **Chemistry**

Is the following sentence true or false?

One potential danger of radon gas is that prolonged exposure to it can lead to lung cancer. true

Detecting Nuclear Radiation

(page 297)16. Name

two devices that are used to detect nuclear radiation.

a. Geiger counters

b. Film badges

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on completing selected  
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approximately 90  
minutes each. Refer to  
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CHAPTER 25, Nuclear  
Chemistry (continued)

Types of Radiation

(pages 800–802) 6.

Complete the following table showing some characteristics of the main types of radiation commonly emitted during radioactive decay. 7. Look at Figure 25.2a on page 801. It shows the alpha decay of uranium-238 to thorium-234. a.

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Problems Section

Review 25.1 Part A

Completion 1

radioactive 2

radioisotopes 3 nuclei

4. ... Nuclear Chemistry

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3.pdf. What students

are saying.

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**chemistry**

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Nuclear Reactions •

Nuclear reactions

involve changes in the

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nucleus, whereas chemical reactions involve the loss, gain, and sharing of electrons. • Different isotopes of the same element may undergo very different nuclear reactions, even though an element's isotopes all share the same chemical characteristics.

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2 16 days 192 g 64 g

25% 3 24 days 224 g

32 g 12.5% 4 32 days

240 g 16 g 6.25% 5 40

days 248 g 8 g 3.125%

6 48 days 252 g 4 g

1.5625% . Half-Life

Problems 1. How much

of a 150g sample of

Au-198 is left after 8.10

minutes if it's half life

is 2.70 minutes? ...

Chapter 25 - Nuclear

Chemistry Author:

Authorized User

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## Chemistry Chapter 25: Nuclear Chemistry ...

Chemistry I Our  
Chemistry I class is an introductory chemistry class, but it may be a challenging class for you at the high school level because it makes you think in ways that you have not had to in other classes before. To be successful you will need to make sure you are prepared everyday to get the

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most out of the class  
this year.

## **Practice Problems (Chapter 10): Nuclear Chemistry**

Chapter 25 Section  
25.2 (continued) Half-  
Life Discuss Explain  
that, for each element,  
there exists only a  
small range of neutron-  
to-proton ratios that  
produce stable nuclei.  
If a nucleus does not  
reflect a stable ratio, it  
spontaneously decays

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until a stable ratio of neutrons to protons results. Relate Explain that the nuclear stability that

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Practice Problems

(Chapter 10): Nuclear  
Chemistry CHEM 30A

1. Write the equation for the nuclear reaction described in each of the following processes: a.

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Americium-241

( $^{241}\text{Am}$ ) undergoes alpha decay (inside a smoke detector) b.

Iodine-131 ( $^{131}\text{I}$ )

undergoes normal beta decay (used in therapy for hyperthyroidism)

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the number of slow

moving neutrons in a

nuclear ...

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692 Chapter 16

Nuclear Chemistry 16.1

The Nucleus and

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Radioactivity Our journey into the center of the atom begins with a brief review. You learned in Chapter 3 that the protons and neutrons in each atom are found in a tiny, central

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Chemistry Section  
10.1 Radioactivity ...**



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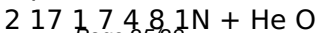
Nuclear Chemistry

Nuclear

Transformations •

Rutherford in 1919 performed the first nuclear transformation.

- The transmutations are sometimes represented by listing in order, the target nucleus, the bombarding particle, the ejecting particle and the product nucleus. • The above equation becomes:



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+ H →  ${}_{14}^{17}$

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... 25.3 Transmutation

pages 815–820

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Chapter 25, 147 Pacing

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LESSON PLAN 25.5

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Nuclear Chemistry

CHAPTER 25 What

You'll Learn You will

trace the history of

nuclear chemistry from

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discovery to application. You will identify types of radioactive decay and solve decay rate problems. You will describe the reactions involved in nuclear fission and fusion.