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Chapter 9: Chemical

Names and Formulas

Section 9., Naming

Ions. The language of

Chemistry.

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### Chemical Names

Chemistry has its own distinct language. With chemical names and formulas, it can all seem a bit daunting. It is important for you to understand how ions and ionic compounds are

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### Chemical Names

the Main Idea As you review each section, write down the key concepts that are found in bold print in each major division.

Then add details and extensions until you have a ... Chemical Names and Formulas

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Assessment

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A B; anion: any atom or  
group of atoms that  
has a negative charge:  
law of multiple

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proportions: Whenever two elements form more than one compound, the different masses of one element that combine with the same mass of the other element are in the ratio of small whole numbers

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Names and Formulas.  
Jennie L. Borders.  
Periodic Table for  
Naming. Section 9.1  
Regular Metals Review.  
The regular metals  
include Group 1  
(except H) and 2 and  
aluminum. When  
naming a compound  
that starts with a  
regular metal, name  
the first element and

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Chemical Names

add -ide to the second element (except for polyatomic ions).

Formulas

Worksheet

## **Chapter 9: Chemical Names and Formulas**

Chapter 11 - Chemical Reactions

Chapter 12 - Stoichiometry

Chapter 13 - States of Matter

Chapter 14 - Behavior of Gases

Chapter 15 - Water and Aqueous Systems

Chapter 16 - Solutions

Chapter 17 - Thermochemistry

Chapter 18 - Reaction

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Chemical Names

Rates and Equilibrium  
... Chapter 9 - Chemical

Names and Formulas.

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Answer Key

**Chapter 9 Chemical**

**Names And**

**Formulas**

**Assessment Answers**

...

The name of a base is  
the name of the cation  
followed by the name  
of the anion

(hydroxide). The  
formula of a base is  
written by showing the  
number of hydroxide

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### Chemical Names

ions needed to balance the positive charge on the cation. After reading Lesson 9.4, answer the following questions. Names and Formulas of Acids hydrogen1.

### **Quia - Chapter 9 "Chemical Names and Formulas"**

Chapter 9 Practice Test  
- Naming and Writing  
Chemical Formulas  
Matching Match each  
item with the correct

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statement below.

Match each item with the correct statement below. a. monatomic ion f. cation b. acid g. binary compound c. base h. anion d. law of definite proportions i. polyatomic ion e. law of multiple proportions

\_\_\_ 1.

**CHAPTER 9 Study Guide - Quia**

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Chemical Formulas

Example 9.3 What is

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## Chemical Names

the empirical formula for a compound having the composition: 49.0% C, 2.7% H and 48.3% Cl? 1. One hundred amu of the compound contains 49.0 amu C, 2.7 amu H and 48.3 amu Cl. 2. Mol values are  $49.0/12.0 = 4.08$  for C,  $2.7/1.0 = 2.7$  for H, and  $48.3/35.5 = 1.36$  for Cl. 3.

## **Chapter 9 - Chemical Names and Formulas**

After you claim an

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answer you'll have 24

hours to send in a

draft. An editor will

review the submission

and either publish your

submission or provide

feedback. Next Answer

Chapter 9 - Chemical

Names and Formulas -

9.1 Naming Ions -

Sample Problem 9.1 -

Page 267: 2 Previous

Answer Chapter 9 ...

**Chapter 9 Chemical**

**Names Formulas**

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**Chapter 9 Practice  
Test - Naming and  
Writing Chemical  
Formulas**

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Names and Formulas 85

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### Chemical Names

Names and Formulas  
for Bases (page 273) 6.

A base is a compound  
that produces \_\_\_\_\_

when dissolved in  
water. 7. How are  
bases named?

SECTION 9.5 THE LAWS  
GOVERNING

FORMULAS AND

NAMES (pages

274–279) This section  
uses data to

demonstrate that a  
compound obeys the  
law of definite

proportions.

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