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Chemistry

Stoichiometry

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6/22/2017 B . Gas

Stoichiometry .

Chemistry 110. 1]

Given the equation: 2

$\text{NH}_3(\text{g}) + 3 \text{Cl}_2(\text{g}) \rightarrow$

$\text{N}_2(\text{g}) + 6 \text{HCl}(\text{g})$. a.

How many milliliters of

nitrogen can be made

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from 13 L of chlorine
and 10.0 L of ammonia
gas at STP?

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MCQ & Answers

While the mole ratio is ever-present in all stoichiometry calculations, amounts of substances in the laboratory are most often measured by mass. Therefore, we need to use mole-mass calculations in combination with mole ratios to solve several different types of mass-based stoichiometry problems.

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12.3: Mass-Mole and

Mole-Mass

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Limiting reagent

stoichiometry. This is

the currently selected

item. Next lesson.

Molecular composition.

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(practice) | Khan

Academy

Mole to Grams, Grams

to Moles Conversions

Worksheet What are

the molecular weights

of the following

compounds? (all

masses must be to

nearest hundredth) 1)

NaOH 2) H_3PO_4 3) H_2O 4)

Mn_2Se_7 5)

$MgCl_2$ 6) $(NH_4)_2SO_4$

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4 There are three definitions (equalities) of mole. They are: 1 mole = 6.02×10^{23} particles

**Mole Calculation
Worksheet -
Brookside High
School**

Introduction to

Stoichiometry

Mastering

Stoichiometry SG 12.2

Limiting Reagents

Limiting Reagents 2

Percent Yield

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Calculations Percent

Yield Lab SG 12.3 &

12.4 Chapter 12

Review Quiz 12.3

Chapter 15 Solubility

Worksheet SG 15.1 &

15.2 Understanding

Molarity Diluting

Solutions Molality &

Percent Solution

Solubility Curve Lab

Electrolytes Lab SG ...

Answer Keys -

HONORS CHEMISTRY

Naming Ionic

Compounds Practice

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Worksheet And

Answers
Worksheet: Write some
names, write some
formulas, repeat. More
Naming Ionic

Compounds : Yes,
you're in luck with
more naming and
formulas. Even More
Naming Ionic

Compounds : If you
liked parts one and two
of the ionic compound
naming saga, you'll be
absolutely thrilled to
practice them some
more.

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Naming Worksheets

| The Cavalcade o'

Chemistry

9.3 Stoichiometry of Gaseous Substances, Mixtures, and Reactions. ... Following are the equations of several nuclear reactions that have important roles in the history of nuclear chemistry: ... Answers to Chemistry End of Chapter Exercises. 1. (a) A nucleon is any particle contained in

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the nucleus of the
atom, so it can refer to
protons and ...

21.2 Nuclear Equations - Chemistry

Example 2. Computing
Formula Mass for an
Ionic Compound

Aluminum sulfate, $\text{Al}_2(\text{SO}_4)_3$, is an ionic
compound that is used
in the manufacture of
paper and in various
water purification
processes. What is the

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formula mass (amu) of
this compound?

Solution The formula
for this compound
indicates it contains Al
3+ and SO 4 2- ions
combined in a 2:3
ratio. For purposes of
computing a formula
mass ...

3.1 Formula Mass and the Mole

Concept - Chemistry

Answer: ACDHIKNO. a.
TRUE - Work is a form
of energy, and in fact it

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Answers

has units of energy.. b. FALSE - Watt is the standard metric unit of power; Joule is the standard metric unit of energy.. c. TRUE - A $N \cdot m$ is equal to a Joule.

d. TRUE - A $kg \cdot m^2 / s^2$ is a mass unit times a speed squared unit, making it a kinetic energy unit and equivalent to a Joule..

e. FALSE - Work is not dependent on ...

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