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CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds.

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1) all matter is made of atoms that can't be divided, created, or destroyed. 2) atoms of an element have identical physical and chemical properties 3) atoms of different elements differ in physical and chemical properties 4) atoms of different elements combine in simple, whole-number ratios, forming compounds 5) In chemical reactions, atoms are combined, separated, or rearranged, but never ...

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