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Write the nuclear symbol for the missing term in equation 5. 139

48Cd 9. Write the nuclear symbol for the missing term in equation 6. 1 1H Name

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WORKSHEET Use with

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Nuclear scattering theory. in the centre of mass frame, where a and b are the initial species about to collide, c is the final species, and R is the resonant state .

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Chapter 11, Section  
11.2 For each of the  
following problems,  
write the balanced  
chemical equation that  
represents the  
reaction. Then  
complete the table  
below by identifying  
the known substance,  
the unknown  
substance, and the  
mole ratio that you  
would use to solve  
each problem

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correctly.

## Balancing Nuclear Science Teaching Transparency Worksheet 15

### Answer ...

in the fission reactions  
in nuclear power plants  
and in the fusion  
reactions in the sun.

This section tells you  
why this energy  
conversion takes place.

Nuclear power is a  
major source of energy  
for electrical  
generation worldwide.

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Nuclear power plants are found in over 30 countries and generate a significant percentage of the world's electricity.

## **Science Teaching Transparency Worksheet 15**

### **Answer ...**

Nuclear Chemistry  
Worksheet K

Directions: Identify the following as alpha, beta, gamma, or neutron. Name: Period:

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56 Date: 231 90

91Kr+3.36 1. 5. 6. 8. 9.

—n Nuclear decay with  
no mass and no charge

An electron Least

penetrating nuclear

decay Most damaging

nuclear decay to the

human body Nuclear

decay that can be

stopped by skin or

paper. 12. 14.

**Name Date Class**

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Equations. You will  
need a periodic table in  
order to complete this  
activity. Choose the  
correct answer for each  
question. Identify the  
missing particle in the  
following nuclear  
reaction: Identify the

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missing particle in the following nuclear reaction: Identify the missing particle in the following nuclear reaction:

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Determining Numbers  
of Ions Use with  
Chapter 7, Section 7.3

1. Write a simple word

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equation that illustrates what must be true of total positive charge and total negative charge in an ionic compound. 2. How many potassium ions (group 1) would be related to balance the charge of each of the following in a compound?

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Chapter 1, Section 11.1

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WORKSHEET 16

Section 11.2 Determine  
the mole ratios for  
each of the following  
balanced chemical

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equations.  $1.2\text{C(s)} + \text{O}_2\text{(g)} \rightarrow 2\text{CO(g)}$  1 mol  $\text{O}_2$   
2 mol C 2 mol  $\text{CO}/2$   
mol C 2 mol C/1 mol  $\text{O}_2$   
2 2 mol  $\text{CO}/1$  mol  $\text{O}_2$   
2 mol C/2 mol  $\text{CO}$  1  
mol  $\text{O}_2$  /2 mol  $\text{CO}$  2 2.  
 $\text{WO}_3\text{(s)} + 3\text{H}_2\text{(g)} \rightarrow \text{W(s)} + 3\text{H}_2\text{O(l)}$

## **Chapter 16 Nuclear Chemistry - Mark Bishop**

ALPHAS, BETAS AND  
GAMMAS OH, MY! 1  
Grade: 8 - 12 Intended  
Learning Outcome(s):

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To develop energy  
literate students.

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Integrated Earth

Systems, Geology,  
Chemistry Materials:

Periodic Table of the  
Elements for each  
student (included)

Decay Event Table  
worksheet (included)

Nuclear Reactions and  
Atomic Notation  
worksheet

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Skills Practiced.

Defining key concepts -  
ensure that you can

accurately define main  
terms, such as beta

decay and gamma

decay. Reading

comprehension -

ensure that you draw

the most important

information from the

related lesson on how

to balance nuclear

equations and predict

the product of a

nuclear reaction.

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