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Solution Molecular Structure Name

CH₃CH₂SH Ethanol CH₃CH₂NH₂

Ethylamine CH₃CH₂CH₂CH₂COOH

Pentanoic acid Cyclopentanol

Cyclopentyl alcohol Hydrocyclopentane

Acetone Propanone Solution $\rho = 0.78$

(100) = (1)(10.73)(90+460) =

(1)(10.73)(90+460) (100) = 59.015 $\times 10^3$

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Therefore, pressure is the main variable that affects the fluid properties and the gas dissolved in oil (solution gas, gas in solution) will come out of solution.

90C Properties Of Petroleum Fluids

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